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Modular Assembly of Axially Chiral QUINAP Derivatives via Nickel-Catalyzed Enantioselective C-P Cross-Coupling

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ABSTRACT: QUINAPs represent a powerful class of ligands with broad applications in a diverse array of synthetically important asymmetric transformations. However, their broader utilization is often constrained by a limited substrate scope and high cost associated with their multistep synthesis. Consequently, a highly efficient dynamic kinetic asymmetric transformation of *N*-heterobiaryl derivatives has been developed through Ni-catalyzed



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asymmetric C-P cross-coupling. This method provides versatile access to a wide range of substituted QUINAP derivatives with good yields and high enantioselectivities, which are successfully applied in both transition-metal catalysis and organocatalysis.

KEYWORDS: Axial Chirality, C-P coupling, Ni-catalysis, QUINAP, DYKAT

A xially chiral phosphine ligands have become fundamentally important in modern organic chemistry due to their pivotal role in catalytic processes, where they significantly influence both catalytic activity and selectivity. Among these, axially chiral *P,N*-ligands, such as QUINAP and Quinazolinap, have emerged as a particularly powerful class, with notable recent advances in their synthesis, resolution, and application to a broad spectrum of synthetically important asymmetric transformations (Scheme 1a). Le, However, the synthesis of axially chiral *P,N*-ligands has predominantly relied on intricate, multistep procedures and resolution methods, which limit the ability to systematically modify the steric and electronic properties of the catalyst.

N-Heterobiaryl derivatives are widely employed in classical dynamic kinetic asymmetric transformation (DYKAT) systems, primarily through the formation of intermediate metallacycles. Over the past decades, noble metal-catalyzed DYKATs, utilizing metals such as Pd, Rh, and Ir, have demonstrated remarkable efficacy in synthesizing axially chiral heterobiaryls.² In 2013, Lassaletta ingeniously developed a dynamic kinetic Suzuki–Miyaura cross-coupling method, employing heterobiaryl triflates and chiral palladium complexes.^{2a} This strategy was subsequently extended to construct a variety of C–C,^{2b–e} C–N,^{2f} and C–P^{2g–i} bonds in a series of axially chiral heterobiaryl compounds. Notably, Rh-catalyzed³ or Ir-catalyzed⁴ C–H bond activation strategies in DYKAT chemistry have also achieved significant advancements, further propelling the development of this field.

In the realm of earth-abundant metal-catalyzed DYKAT chemistry, the Tang group recently achieved notable break-throughs, including a photomediated cobalt-catalyzed asymmetric radical coupling reaction and 1,4-addition reactions. Meanwhile, Wang reported a significant advancement with the development of a cobalt-catalyzed asymmetric reductive

alkenylation and arylation of heterobiaryl tosylates.⁶ More recently, Liang introduced a groundbreaking dynamic kinetic reductive Grignard-type addition for the simultaneous construction of axial and central chirality through Cocatalysis.⁷ Additionally, Cao developed a pioneering nickel-catalyzed enantio-convergent transformation of *N*-heterobiaryl derivatives using Grignard reagents.⁸ Lately, Yu reported an innovative nickel-catalyzed enantio-convergent carboxylation of aza-biaryl triflates with CO₂, enabling the synthesis of atropisomeric carboxylic acids.⁹ Despite these advancements, there have been no reported examples of asymmetric C–P bond construction for axially chiral *P*,*N*-ligands through Ni catalysis.

Asymmetric C–P bond construction represents a versatile and practical strategy for synthesizing axially chiral phosphorus compounds. This includes methodologies such as asymmetric C–P coupling, allenylation of phosphorus nucleophiles, addition of phosphorus nucleophiles to alkynes, and ringopening reactions. Despite the development of numerous methods for the preparation of axially chiral phosphorus compounds, the direct construction of C–P bonds through transition-metal-catalyzed approaches to synthesize axially chiral *P*,*N*-ligands remains a significant challenge. This challenge likely stems from the strong coordination between phosphorus nucleophiles and metal catalysts. The asymmetric synthesis of chiral QUINAP was not achieved until 2013, when

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Scheme 1. Ni-Catalyzed Asymmetric C-P Coupling for Synthesis of Chiral QUINAP Derivatives

a. Representative axial chiral P-N ligands

b. Synthesis of axially chiral P-N ligand via asymmetirc C-P coupling

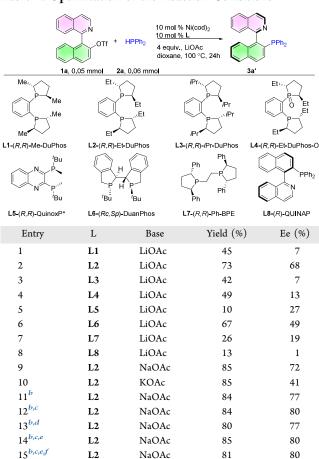
c. This work: enantioselective synthesis of QUINAP derivatives via Ni-catalyzed C-P coupling

Stoltz and Virgil reported the kinetic resolution of aryl bromides and dynamic kinetic resolution of aryl-triflates asymmetric using Pd-catalyzed C–P coupling. Subsequent research has focused on synthesizing axially chiral P, N ligands via DKR process involving C–P coupling with aryl triflates (Scheme 1b). However, these methods suffer from significant limitations, including poor substrate generality, the reliance on expensive masked phosphine sources (TMSPAr₂), and the need for prolonged addition of HPAr₂ to minimize coordination between phosphine and metal catalysts. Additionally, the Tan group developed a kinetic resolution method for synthesis of axially chiral QUINAP derivatives through ketone-catalyzed enantioselective oxidation. ¹³

Herein, building on traditional Ni-catalyzed C-P cross-coupling ¹⁴ and our group's previous work in chiral phosphorus chemistry, ^{10d,11b,15} we have developed a Ni-catalyzed DYKAT for the C-P coupling of *N*-heterobiaryl triflates and HPAr₂. This method provides a highly efficient approach for the structurally diverse construction of chiral QUINAP derivatives in high yields with excellent enantioselectivities (Scheme 1c).

We initially achieved Ni-catalyzed asymmetric C-P cross-coupling through dynamic kinetic asymmetric transformations using N-heterobiaryl triflate (1a) and HPPh₂ (2a) as model substrates in the presence of Ni(cod)₂ and chiral phosphine ligands (L1-L8) (Table 1). After optimization of the chiral ligands, we found that (R,R)-Et-DuPhos (L2) gave the best enantioselectivity (entries 1–8). (R,R)-Et-DuPhos-O (L4) was able to catalyze the reaction with a suboptimal result. Interestingly, (R)-QUINAP (L8) afforded 13% yield and 1% ee, suggesting that the product 3a' can act as a ligand to some extent, thereby influencing enantioselectivity. This observation explains the challenges of achieving higher enantioselectivity. Then, using NaOAc instead of LiOAc increased the

Table 1. Optimization of the Reaction Conditions^a

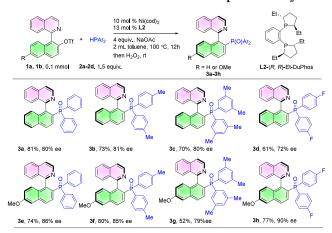


^aReaction conditions: **1a** (0.05 mmol), **2a** (0.06 mmol), 10 mol % Ni(cod)₂, 10 mol % **L**, in 1 mL of dioxane, 100 °C, 24 h. Yield (**3a**') was based on ³¹P NMR analysis with PPh₃ as internal standard. Ee was determined by HPLC analysis. ^bToluene was used instead of dioxane. ^c13 mol % **L2**, 12 h. ^d5 mol % Ni(cod)₂, 6.5 mol % **L2**, 12 h. ^e0.075 mmol **2a**. ^fA scale of 0.1 mmol. Then 4 equiv of H_2O_2 was added. Isolated yield of QUINAPO **3a**.

enantioselectivity to 72% and KOAc gave poor enantioselectivity (entries 2, 9, and 10). Toluene outperformed 1,4-dioxane (entry 9 vs 11). Reducing the reaction time and increasing the loading of L2 enhanced the enantioselectivity to 80% (entry 12). However, reducing the loading of Ni(cod)₂ and L2 led to a decrease in the enantioselectivity (entry 13). The yield improved slightly when the amount of 2a was increased to 1.5 equiv (entry 14). Finally, under the optimized conditions at a scale of 0.1 mmol and further oxidation by $\rm H_2O_2$, QUINAPO 3a was obtained in 81% isolated yield and 80% ee (entry 15). The configuration of the product (3a') was consistent with that of commercially available (R)-QUINAP.

Having established the optimal conditions for this dynamic kinetic asymmetric C-P coupling, we initially explored various HPAr₂ (Table 2 2a-2d) with 1a, yielding promising results (3a-3d, 61-81% yield, 72-81% ee). Furthermore, 6-MeOsubstituted N-heterobiaryl triflates (1b) reacted with different HPAr₂ (2) to give improved yields and ee values (3e-3h, 52-80% yield, 79-90% ee) compared to 1a. Various substituents on the naphthalene ring of N-heterobiaryl triflates were also investigated (Table 3). The presence of an ester group led to a decrease in enantioselectivity (3l, 79% yield, 75% ee), while the electron donating group enhanced enantioselectivity (3i-

Table 2. Evaluation of the Substrate Scope of HPAr,

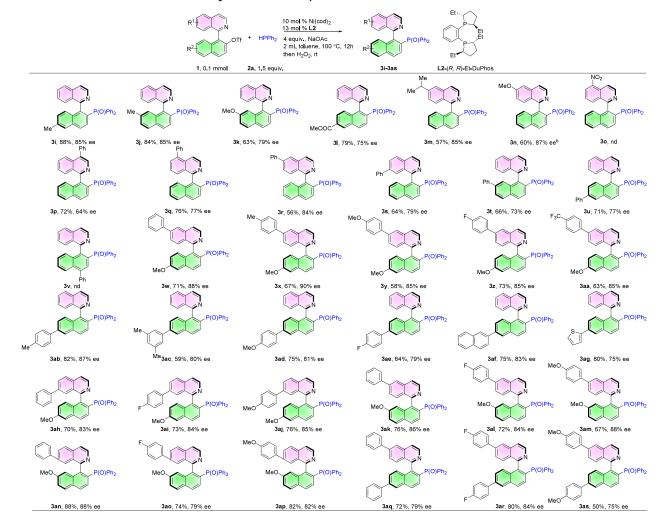


"Reaction conditions: 1a (R = H) or 1b (R = OMe) (0.1 mmol), 2a-2d (0.15 mmol), 10 mol % Ni(cod)₂, 13 mol % L2 in 2 mL of toluene, 100 °C, 12 h. Then 4 equiv of H_2O_2 was added. Isolated yield. Ee was determined by HPLC analysis.

3k, 63–88% yield, 79%–85% ee). 6-Substituent isoquinoline-based *N*-heterobiaryl triflates were also compatible with the reaction, although the electron-donating group slightly reduced yields (**3m** and **3n**, 57–60% yield, 85–87% ee). A strong electron withdrawing group ($-NO_2$, **3o**) resulted in a complex reactivity.

Next, the introduction of a phenyl group to the naphthalene or isoquinoline ring resulted in good yield and enantioselectivity (Table 3, 3p-3u, 56-76% yield, 64-84% ee). However, 4-Ph-naphthalene triflates failed to react, likely due to steric hindrance (3v). To further explore the scope of the reaction, we selected 6-MeO-naphthalene as a constant and examined the effect of various substituents on the isoquinoline ring. Both electron-donating groups (Me and OMe) and electron-withdrawing groups (F and CF₃) were well-tolerated, yielding satisfactory results (3w-3aa, 58-73% yield, 85-90% ee). Moreover, N-heterobiaryl triflates bearing electrondonating groups (Me and OMe), electron-withdrawing groups (F), as well as 2-naphthalene or thiophene on the naphthalene ring, also produced QUINAPOs efficiently (3ab-3ag, 59-82% yield, 75-87% ee). Furthermore, different -Ar groups and -OMe positioned at various locations on N-heterobiaryl triflates also showed good practicality (3ah-3ap, 67-88%

Table 3. Evaluation of the Substrate Scope of Heterobiaryl Triflates with Different Substituents^a

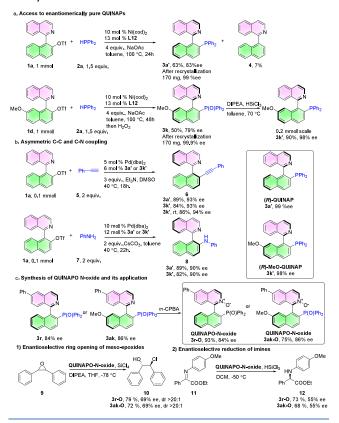


[&]quot;Reaction conditions: 4 (0.1 mmol), 2a (0.15 mmol), 10 mol % Ni(cod)₂, 13 mol % L2 in 2 mL of toluene, 100 °C, 12 h. Then 4 equiv of H₂O₂ was added. Isolated yield. Ee was determined by HPLC analysis. ^b24 h.

yield, 79–88% ee). Finally, substrates featuring 6-Arnaphthalene and 6-Ar-isoquinoline also smoothly yielded the desired products (3aq–3as, 50–80% yield, 75–84% ee).

To evaluate the efficiency of the current method, a 1 mmol scale reaction was conducted, yielding 3a' (63%, 83% ee) and 3k (50%, 79% ee). During the scale-up reaction, we observed detriflate and protonation products (4), which are common side products in transition metal-catalyzed C-P coupling reactions. ¹⁶ Products (3a' and 3k) with 99% ee were obtained after recrystallization. Compound 3k was further transformed via reduction with HSiCl₃, affording 3k' in 90% yield with 98% ee (Scheme 2a). (R)-MeO-QUINAP (3k') demonstrated

Scheme 2. Access to Enantiomerically Pure QUINAPs and Applications in Enantioselective and Diastereoselective Catalysis



catalytic activity comparable to that of (*R*)-QUINAP (3a') in asymmetric C-C and C-N coupling reactions, showcasing excellent catalytic efficiency (Scheme 2b). Additionally, compounds (3r and 3ak) were oxidized by *m*-CPBA to synthesize (*R*)-QUINAPO *N*-oxide 3r-O and 3ak-O in good yields without loss of enantioselectivity, which shows potential application in enantioselective and diastereoselective ring opening of meso-epoxides and reduction of imines in good yields and moderate enantioselectivities without condition optimization (Scheme 2c).

To gain insight into the nickel-catalyzed asymmetric C-P coupling reaction, control experiments were conducted. The yield and enantioselectivity decreased significantly in the absence of a nitrogen atom (15, 43%, 22% ee), suggesting that the formation of a five-membered nickelacycle is essential for this asymmetric transformation. The quinazoline skeleton gave racemic product (16, 87% yield, racemic), likely due to the weak coordination of the nitrogen atom in quinazoline to

the Ni catalysis and its quickly racemization in high temperature (see the Supporting Information) (Scheme 3a).

Scheme 3. Mechanistic Investigations

Based on recent work about asymmetric synthesis of *N*-heterobiaryl derivatives via Ni-catalysis^{8,9} and the control experiment, a possible mechanism is proposed as follows. First, the coordination of the nitrogen atom in isoquinoline to the generated Ni⁽⁰⁾L2 complex facilitates the oxidative addition of the C–O bond in triflates (1a) to Ni⁽⁰⁾L2, forming 5-membered cationic nickelacycle (Ar–Ni^(II)L2) diastereoisomers I and II, which was the enantiodetermining step. These intermediates are prone to epimerization.¹⁷ Following a ligand exchange step, intermediate III is generated, which undergoes competitive protonolysis of the Ni–C bond within the biaryl entity. This protonolysis leads to the formation of a byproduct (4). The primary pathway involves a reductive elimination, yielding 3a' while regenerating the Ni⁽⁰⁾L2 complex (Scheme 3b).

In summary, we have developed an atroposelective DYKATs strategy for the asymmetric synthesis of QUINAP derivatives by using a Ni/(R,R)-Et-DuPhos catalyzed C-P cross-coupling approach. A wide range of N-heterobiaryl triflates and HPAr₂ participated efficiently in the reaction, generating chiral QUINAPs and QUINAPOs in high yield (up to 88%) and enantioselectivities (up to 90% ee). Moreover, we demonstrated the utility of these QUINAP derivatives in asymmetric metal catalysis and organocatalysis, exhibiting good catalytic activity. Future studies will focus on expanding the QUINAP ligand toolbox to address challenging problems in asymmetric catalysis.

ASSOCIATED CONTENT

Solution Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acscatal.5c01663.

Experimental details and procedures, characterization data, HPLC chromatograms and NMR spectra of isolated compounds (PDF)

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Author Contributions

Dr. Z. Yang conceived the project, performed the experiments, and prepared the Supporting Information. Mr. X. Gu repeated some experiments and collected some data. Prof. L.-B. Han contributed to the mechanism study. Prof. J. Wang directed the project. Dr. Z. Yang and Prof. J. Wang wrote the paper with input from all other authors.

Notes

The authors declare no competing financial interest.

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