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NiH-Catalyzed Homobenzylic Hydroalkylation of Aryl Alkenes Using Sulfoxonium Ylides

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ABSTRACT: We present here the first example of NiH-catalyzed homobenzylic hydroalkylation of aryl alkenes mediated by a nickel carbene radical, achieving the elusive β -selectivity with excellent regiocontrol. Mechanistic investigations suggest that this transformation is enabled by the preferential engagement of NiH with bench-stable sulfoxonium ylides, whose unique chelation properties promote carbene activation prior to alkene insertion. The resulting nickel carbene radical is proposed to undergo selective β -addition, followed by intramolecular metal hydride transfer and protodemetalation. The reaction exhibits broad scope across aryl, heteroaryl, and complex bioactive alkene derivatives, as well as

conventional selectivities

R

Conventional selectivities

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Conventional selectivities

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Crucial role of sulfoxonium ylide

Carbene radical

diverse sulfoxonium ylides. This work establishes a new mechanistic platform for NiH catalysis, expanding the synthetic repertoire for site-selective alkene functionalization.

■ INTRODUCTION

The development of synthetic strategies to diversify molecular frameworks through regioselective hydrofunctionalization of alkenes has been a longstanding objective in synthetic chemistry, as such transformations enhance both the synthetic utility and structural complexity of ubiquitous olefin substrates. 1-11 In this context, transition metal hydridecatalyzed hydroalkylation of aryl-containing alkenes has been actively explored as a powerful strategy to access structurally valuable sp³-rich fragments (Scheme 1a, left). 12-16 Among the notable advances, alkylation at the benzylic position has been the most prevalent. 17-21 Elegant studies have shown that in situ generated NiH species, derived from nickel salts and hydride donors, can catalyze formal hydroalkylation of vinylarenes at the benzylic position through umpolung strategies employing electrophilic alkyl sources. In parallel, terminal alkylation has been achieved via olefin chain-walking under redox neutral conditions.8 This approach to remote formal hydroalkylation, particularly with alkenes bearing aryl groups, has been independently demonstrated by the Zhu, 22,23 Martin, 24,25 Qi,²⁶ and Shang²⁷ groups.

Despite the substantial body of these works, regioselective internal alkene hydroalkylation at the *homobenzylic* site remains elusive (Scheme 1a, right). This scarcity is likely attributable to two intrinsic features of transition metal hydrides: the kinetic preference for hydrometalation at the α -position^{28–31} as well as thermodynamic preference of the metal center for the benzylic^{28,29,32–37} or terminal positions, ^{34,37–39} where chain walking is more feasible. Indeed, beyond hydroalkylation across the broader field of hydrofunctionalization, only a handful of examples exhibiting homobenzylic selectivity have

been reported. For instance, metal hydride-mediated hydroborylation at this position was independently demonstrated by the Koh⁴⁰ and Zhang⁴¹ groups. Our group has also contributed to this area by developing a formal NiH-catalyzed enantioselective hydroamidation of vinylarenes, in which C–N bond formation occurs at the homobenzylic position via olefin insertion into a proposed amidonickel intermediate.⁴²

An emerging strategy for hydroalkylation leverages transition metal hydrides in conjunction with inner-sphere carbenoid transfer. A3-45 Notably, our group reported a regioselective remote C—H alkylation of internal olefins via iridium hydride catalysis, combining olefin chain-walking with carbenoid migratory insertion from sulfoxonium ylides (Scheme 1b, left). In contrast, Gevorgyan and co-workers more recently demonstrated that nickel catalysis can reverse the regioselectivity toward the benzylic position using diazo compounds (Scheme 1b, right). In both cases, the observed selectivities reflect the intrinsic behaviors of metal hydrides described above, leaving alternative regioselectivities largely unexplored.

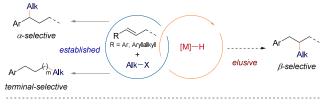
To this end, we envisioned developing a new hydroalkylation platform for aryl alkenes that will deliver homobenzylic selectivity by harnessing carbenoid transfer. A major hurdle in this realm is the inherent propensity of

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Scheme 1. Transition Metal Hydride-Catalyzed Hydroalkylation of Aryl Alkenes

(a) Hydroalkylation of aryl-containing alkenes using alkyl electrophiles



(b) Use of carbene under transition metal hydride catalysis

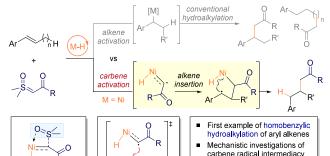
chelation promoted

carbene formation

(c) This work: transposed sequence of carbene activation via NiH catalysis

selectivity determined

by radical stabilization



Broad substrate scope over

Robust carbene precursor

(hetero)aryl alkenes

transition metal hydrides to undergo rapid hydrometalation with alkenes, leading to the more conventional hydroalkylation products (Scheme 1c, top). 43,44 An alternative strategy would selectively activate the carbene precursor prior to alkene engagement, thereby enabling subsequent carbene insertion at the homobenzylic position. Crucially, accessing a carbene species with radical character would select the bond-forming carbon being controlled by radical stabilization, 46-49 thus making homobenzylic carbene insertion highly predictable. While metal carbene radicals have recently emerged as powerful intermediates for C-C bond-forming reactions, we posited that our previously developed NiH systems, 51-54 capable of nitrenoid activation and engaging in paramagnetic species, would engender controlled access to carbene radical intermediates. Furthermore, the stepwise radical addition of a carbenoid via NiH catalysis is anticipated to proceed with suppression of undesired carbene reactivities such as olefin cyclopropanation⁵⁵⁻⁷² and hydride-mediated carbene reducsince intramolecular NiH-mediated hydride transfer would be highly facile under these conditions. 77-80

Herein, we report the first example of NiH-catalyzed homobenzylic hydroalkylation of aryl alkenes using convenient, bench-stable sulfoxonium ylides 81-86 as key carbene precursors (Scheme 1c, bottom). Combined experimental and computational studies reveal that NiH species preferentially engage the carbene precursor first, crucially facilitated by chelation, to generate nickel carbene radicals. This intermediate subsequently undergoes regioselective olefin radical addition, intramolecular metal hydride transfer, and protodemetalation.

RESULTS AND DISCUSSION

Initial Investigation. We commenced our study of olefin hydroalkylation using (E)-1-methoxy-4-(prop-1-en-1-yl)benzene (1) as a model substrate in reactions with various carbene precursors, aiming to identify a suitable precursor for the transposed carbenoid activation (Scheme 2). For the initial

Scheme 2. Initial Investigation on Carbene Precursors^a

^aYields determined by ¹H NMR using 1,1,2-trichloroethane as an internal standard; PMP, para-methoxyphenyl; Ar, 3-fluorophenyl.

screening, we employed a well-defined NiH catalytic system comprising NiBr2·diglyme and bioxazoline ligand L1, in combination with HBpin as the hydride source. Additives were included to promote reactivity: LiCl to activate the reaction components and H₂O as a proton source.

A survey of carbene precursors revealed that representative diazo compounds 46,47,\$5,56,58,61,67,72,75 (C1, C2) failed to produce any alkylated product under the initial conditions. Likewise, other common carbene precursors, including hydrazone 49,59,60,68 (C3), gem-dibromide 62-66,69,70 (C4), and diazirine⁸⁷ (C5), were unreactive. Gratifyingly, when sulfoxonium ylide $^{81-86}$ (C6) was examined, the β -alkylated product was obtained as a single regioisomer in 20% yield. Importantly, no isomeric α -alkylated product was detected in this transformation. Sulfoxonium ylides are attractive surrogates for α -diazocarbonyl compounds: they are typically benchstable, readily prepared from the corresponding carboxylic acids, and compatible with a broad range of reaction conditions. 81-86

Along with the modest yield of the desired product from the reaction of alkene 1 and sulfoxonium ylide C6, several side products arising from carbene reactivity were also observed. These side products included the protonated acetophenone **B1** (25%) and the carbene dimeric adduct **B2** (17%). In addition, saturated alkane B3 was obtained in 65% yield, presumably originating from nickel hydride reactivity. In fact, protonation at the carbene carbon to form a metal-alkyl species is well documented, as evidenced by Fogg and Baker, who showed that hydrated carbene species can undergo dimerization via alkyl-alkyl coupling.⁸⁸ Consistent with this precedent, when C6 was subjected to the NiH reaction conditions in the absence of the alkene substrate, the C-C dimerization product B2 was formed (63%), supporting the potential involvement of a Ni-carbene intermediacy under these conditions (see Supporting Information for details).

Reaction Optimization. With the promising initial results in hand, we carried out further optimization studies using aryl alkene 1 and sulfoxonium ylide C6 (2) as shown in Table 1.

Table 1. Reaction Optimization^a

2 HCO ₂ H/Et ₃ N instead of HBpin <1 3 (MeO) ₃ SiH instead of HBpin 67 4 L3 instead of L2 54 5 L4 instead of L2 75 6 L5 instead of L2 74 7 w/o L2 <1 8 w/o LiCl 52 9 CsF instead of LiCl <1 10 K ₂ CO ₃ instead of LiCl 21 11 'PrOH instead of H ₂ O 70 12 PhCl as solvent 59 13 PhMe as solvent 34 14 w/o NiBr ₂ -diglyme <1 15 w/o H ₂ O 21	entry	deviation from the reaction conditions	3 (%)
3 (MeO) ₃ SiH instead of HBpin 67 4 L3 instead of L2 54 5 L4 instead of L2 75 6 L5 instead of L2 74 7 w/o L2 <1	1	none	79 (76 ^b)
4	2	HCO ₂ H/Et ₃ N instead of HBpin	<1
5 L4 instead of L2 75 6 L5 instead of L2 74 7 w/o L2 <1	3	(MeO) ₃ SiH instead of HBpin	67
6 L5 instead of L2 74 7 w/o L2 <1	4	L3 instead of L2	54
7 w/o L2 <1	5	L4 instead of L2	75
8 w/o LiCI 52 9 CsF instead of LiCI <1	6	L5 instead of L2	74
9 CsF instead of LiCl <1 10 K ₂ CO ₃ instead of LiCl 21 11 'PrOH instead of H ₂ O 70 12 PhCl as solvent 59 13 PhMe as solvent 34 14 w/o NiBr ₂ -diglyme <1 15 w/o H ₂ O 21	7	w/o L2	<1
10 K_2CO_3 instead of LiCl 21 11 $^{'}$ PrOH instead of H_2O 70 12 PhCl as solvent 59 13 PhMe as solvent 34 14 W/O NiBr ₂ -diglyme <1	8	w/o LiCI	52
11 'PrOH instead of H ₂ O 70 12 PhCl as solvent 59 13 PhMe as solvent 34 14 w/o NiBr ₂ ·diglyme <1	9	CsF instead of LiCI	<1
12 PhCI as solvent 59 13 PhMe as solvent 34 14 w/o NiBr ₂ ·diglyme <1	10	K ₂ CO ₃ instead of LiCI	21
13 PhMe as solvent 34 14 w/o NiBr ₂ ·diglyme <1	11	ⁱ PrOH instead of H ₂ O	70
14 w/o NiBr ₂ -diglyme <1 15 w/o H ₂ O 21	12	PhCI as solvent	59
15 W/o H ₂ O 21	13	PhMe as solvent	34
2	14	w/o NiBr₂·diglyme	<1
16 w/o HBpin <1	15	w/o H ₂ O	21
	16	w/o HBpin	<1

^aReaction conditions: 1 (0.1 mmol), 2 (2.0 equiv), NiBr₂-diglyme (10 mol %), L2 (15 mol %), HBpin (2.0 equiv), LiCl (1.0 equiv), H_2O (2.0 equiv), and PhCl/PhMe (7:3, 1.0 mL) at 25 °C for 12 h under N_2 atmosphere; >20:1 regioselectivity was obtained; yields determined by ¹H NMR using 1,1,2-trichloroethane as an internal standard. ^bIsolated yield.

After extensive screening of reaction parameters (see Supporting Information for details), the β -selective hydroalkylation product 3 was obtained in 79% yield with excellent regioselectivity (>20:1) when employing NiBr₂·diglyme (10 mol %) supported by bisoxazoline ligand L2 (15 mol %), in the presence of HBpin (2.0 equiv) as the hydride source, LiCl (1.0 equiv) as an additive, and H₂O (2.0 equiv) as the proton source, at 25 °C in a PhCl/PhMe (7:3) cosolvent system (entry 1).

Changing the hydride source to HCO₂H/Et₃N completely suppressed product formation (entry 2), whereas using (MeO)₃SiH instead afforded 67% yield of 3 (entry 3). Additional modifications to the ligand structure failed to improve the yield (entries 4–6; see Supporting Information for further details), and no reaction occurred in the absence of ligand (entry 7). Given that sulfoxonium ylides can be converted to carbenes more effectively in the presence of certain additives, 43,82,84-86 we next examined alternative species. In the absence of LiCl, the yield dropped to 52% (entry 8), and other additives such as CsF and K₂CO₃ proved less effective (entries 9 and 10, respectively). Notably, replacing water with isopropanol as the proton source delivered a comparable yield (entry 11). Using either PhCl or PhMe as a single solvent resulted in diminished yields (entries 12 and 13). Control experiments confirmed that the reaction required both the nickel catalyst and HBpin, and that

water as proton source was essential for efficient conversion (entries 14-16).

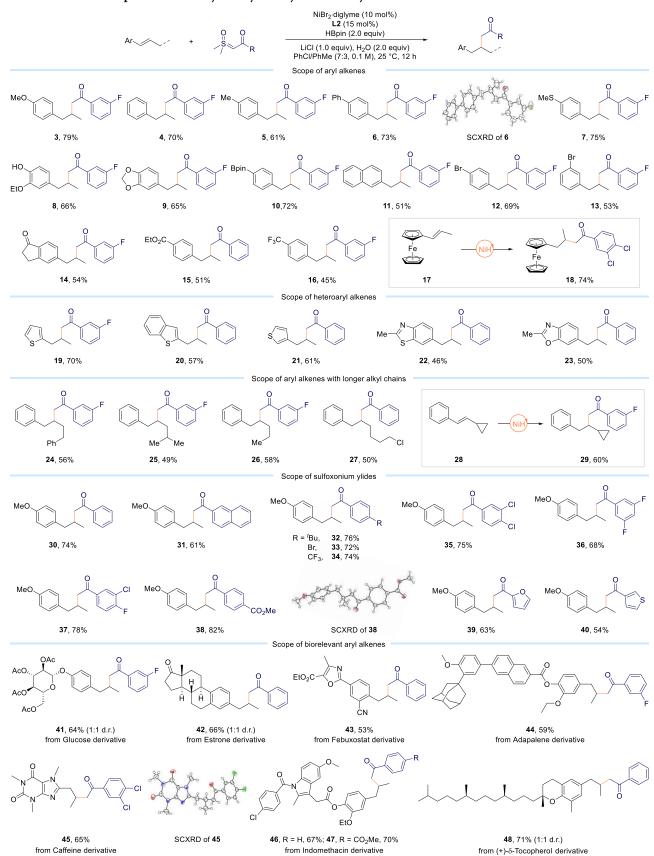
Reaction Scope. Having established the optimized reaction conditions, we next explored the scope of the regioselective homobenzylic hydroalkylation of aryl alkenes (Scheme 3). All substrates tested delivered excellent regioselectivity (>20:1), favoring hydroalkylation at the homobenzylic position over both benzylic hydroalkylation and olefin cyclopropanation pathways. Vinylarenes bearing electron-donating or neutral substituents on the aryl ring, including alkoxy (3, 9), hydrogen (4), methyl (5), phenyl (6), methylthio (7), hydroxy (8), and Bpin (10) groups, were well tolerated, affording the corresponding ketone products in high yields. The structure of product 6 was unambiguously confirmed by single-crystal X-ray diffraction (SCXRD) analysis. Naphthyl-substituted olefins also furnished the desired product in moderate yield (11). Electron-withdrawing substituents, such as bromo (12, 13), acyl (14), ester (15), and trifluoromethyl (16) groups, were likewise compatible, delivering the corresponding products in moderate to good yields. Interestingly, a ferrocene derivative 17, representing a distinct class of organometallic framework, also participated successfully to give product 18 in good yield with high regioselectivity.

Furthermore, heteroaryl alkenes featuring medicinally relevant scaffolds, 89,90 including thiophenes (19, 21), benzothiophene (20), benzothiazole (22), and benzoxazole (23), underwent smooth hydroalkylation to furnish the corresponding products with consistently high regioselectivity. Vinylarenes bearing extended alkyl chains were also amenable to the transformation, affording the corresponding products in moderate yields with excellent regioselectivity. Indeed, substrates containing alkyl chains with additional functional groups, such as phenyl (24), alkyl (25, 26), and chloro (27) substituents, were well tolerated. Notably, vinylarenes bearing a cyclopropyl group at the β -position underwent alkylation smoothly, delivering the desired products without opening of the cyclopropane ring (29).

As sulfoxonium ylides are readily prepared from the corresponding carboxylic acids in two steps and in high yields, we next examined the scope of these carbene precursors by employing various derivatives (30–40). Benzoyl sulfoxonium ylides bearing diverse substituents on the aryl ring, such as benzofused (31), alkyl (32), trifluoromethyl (34), halides (33, 35–37), and ester (38) groups, were well accommodated, affording the corresponding ketone products in high yields with excellent regioselectivity. Furthermore, sulfoxonium ylides tethered to heterocycles, such as furan (39) and thiophene (40), were also tolerable under the optimized conditions. However, the reaction was ineffective with sulfoxonium ylides bearing aliphatic acyl groups (see Supporting Information for details on unsuccessful substrates).

Next, we evaluated the applicability of the present homobenzylic hydroal kylation approach to alkene substrates derived from complex or biologically relevant molecules (41–48). Significantly, a range of ketone analogues of drugs and natural products were obtained in high yields and with excellent regioselectivity, including derivatives of glucose (41), estrone (42), febuxostat (43), adapalene (44), caffeine (45), indomethacin (46, 47), and (+)- δ -tocopherol (48), thereby demonstrating the high synthetic utility of the current homobenzylic hydroal kylation. The structure of the ketone

Scheme 3. Reaction Scope of NiH-Catalyzed Hydroalkylation of Aryl Alkenes^a



[&]quot;Reaction conditions: alkene (0.1 mmol), sulfoxonium ylide (2.0 equiv), NiBr₂-diglyme (10 mol %), **L2** (15 mol %), HBpin (2.0 equiv), LiCl (1.0 equiv), H₂O (2.0 equiv), and PhCl/PhMe (7:3, 1.0 mL) at 25 °C for 12 h under N₂ atmosphere; isolated yields are reported; >20:1 regioselectivity was obtained.

product obtained from the caffeine derivative (45) was further confirmed by SCXRD analysis.

Mechanistic Investigations. With the broad generality of the NiH-catalyzed homobenzylic hydroalkylation, we next wished to substantiate our proposed mechanistic blueprint. For this investigation, we first performed density functional theory (DFT) calculations based on a catalytic cycle in which Ni(I)—H serves as the active species (Figure 1a), guided by previously

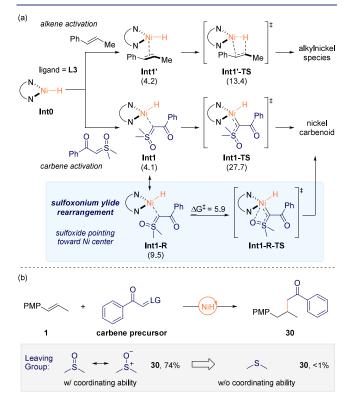


Figure 1. (a) Proposed initial activation modes by NiH, and related energies calculated at the M06/def2-TZVPP+SMD(PhCl)//M06/def2-SVP level of theory (ΔG in kcal/mol). (b) Dependency of carbene precursor leaving group on the reactivity.

reported mechanisms for NiH-catalyzed hydrofunctionalization reactions. $^{21,22,32-35,39,42,51-54}$ In fact, consistent with precedents, the calculated energy barrier for direct hydrometalation of NiH into the alkene (Int1'-TS, 13.4 kcal/mol) was markedly lower than that for nickel carbenoid formation (Int1-TS, 27.7 kcal/mol), favoring a benzylnickel intermediate, in line with the regioselectivity typically observed from NiH insertion into vinylarenes. 20,44 However, in light of the critical role of the sulfoxonium ylide as the carbene precursor, our computational analysis suggests that it may possess a unique chelating property: upon rotation of the sulfoxide moiety toward the nickel center, the sulfoxide oxygen can stabilize the otherwise high-energy transition state for carbene formation. This chelation-promoted carbene activation was predicted to proceed with an energy barrier of 5.9 kcal/mol, indicating that prior activation of the sulfoxonium ylide is also feasible under the current mild conditions. Although activation of either olefin or sulfoxonium ylide was estimated to occur readily, we propose that the excess equivalents and zwitterionic nature of the latter will render nickel carbene formation to be

more probable and irreversible. Indeed, multiple literature precedents have documented the coordinating ability of sulfoxonium ylides through their innate resonance-stabilized donor sites. $^{83,92-94}$

To support this proposition, we examined the dependence of the carbene precursor's nature on the desired reactivity (Figure 1b). As with the other well-known carbene precursors that were unsuccessful (Scheme 2), removal of the oxygen atom from the sulfoxide moiety completely suppressed the current reaction. Considering that sulfonium ylides are also frequently employed as carbene precursors, 81–83,86 this leaving-group dependence further supports the importance of a plausible coordination-assisted carbene activation, which indeed enables the transposed sequence.

Having pinpointed the Ni-carbenoid intermediate with radical character as an essential mechanistic element that enables the key regioselective carbene insertion, we sought to experimentally validate this radical nature. Initially, introduction of 2,2,6,6-tetramethylpiperidinyloxy (TEMPO) under the standard reaction conditions completely suppressed the formation of the desired product, indicating that the reaction likely proceeds via a radical pathway (Figure 2a). Furthermore,

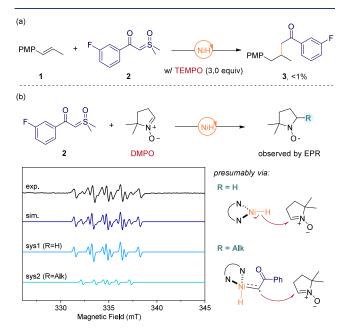


Figure 2. (a) Radical-trap experiment using TEMPO. (b) Experimental EPR spectra of spin-trapped intermediates; simulation parameters for sys1: $g_{\rm iso} = 2.007$, $a^{\rm N} = 42$ MHz, $a^{\rm H} = 52$ MHz, $a^{\rm H} = 52$ MHz; simulation parameters for sys2: $g_{\rm iso} = 2.007$, $a^{\rm N} = 38$ MHz, $a^{\rm H} = 65$ MHz.

spin-trapping experiments with 5,5-dimethyl-1-pyrroline *N*-oxide (DMPO) were performed to detect radical-trapped intermediates by electron paramagnetic resonance (EPR) spectroscopy (Figure 2b). In this case, multiple species were observed, consistent with trapping of a hydride-derived species from NiH or formation of an alkyl-trapped species via the carbene radical. These results corroborate earlier EPR detections of spin-trapped carbene radical species ^{59,95,96} or analogous intermediates in NiH systems involving alkyl halides. ⁹⁷

Scheme 4. Mechanistic Investigations on Nickel Carbenoid Insertion^a

"(a) Computed pathways of nickel carbenoid insertion and subsequent hydride transfer calculated at the M06/def2-TZVPP+SMD(PhCl)//M06/def2-SVP level of theory (ΔG in kcal/mol); MA, methyl acetate. (b) Radical clock experiment. (c) Investigation on involvement of cyclopropane ring-opening. (d) Deuterium experiments.

On the basis of the proposed nickel carbenoid intermediate possessing radical character, we next scrutinized the subsequent carbene insertion step, where the regioselectivity is established (Scheme 4a). DFT calculations revealed that the nickel carbenoid intermediate Int2 indeed exhibits radical character, with a spin density of -0.383 on the carbene carbon. Radical insertion of Int2 into the olefin was calculated to be feasible (Int2-TS), with an energy barrier of 14.7 kcal/mol. The resulting Ni–alkyl species Int3 displays a spin density of 0.728 on the benzylic carbon, indicating that the observed regioselectivity arises from radical stabilization at the benzylic position as expected. Subsequently, the metal hydride transfer process of Ni–H is predicted to occur rapidly, owing to its small energy barrier (Int3-TS, 4.5 kcal/mol), to furnish the hydroalkylated nickel species Int4.

Providing additional evidence for the regioselectivitydetermining step of the nickel carbene radical addition (Int2-TS), we sought to examine whether certain chiral ligands may induce enantioselectivity at the homobenzylic carbon center (Scheme 4a, bottom). To evaluate this plausibility, a series of chiral ligands were screened using alkene 1 and sulfoxonium ylide 2 as model substrates under slightly modified conditions. The standard bisoxazoline ligand L2 provided the desired product in 65% yield with 30% ee, whereas ligand L6 improved enantioselectivity (90:10 e.r.), but giving rise to low reactivity (28%). These preliminary results support our proposal that olefin insertion of the putative nickel carbene radical intermediate governs both regio- and enantioselectivity. On the other hand, to exclude an alternative pathway of involving cyclopropanation and subsequent ringopening, when cyclopropane 49 was subjected to the standard conditions, no desired product was detected (Scheme 4b). This observation suggests that the conventional cyclopropanation process is not operative in our present system.

The above-mentioned reaction pathway was corroborated by experimental mechanistic studies. For instance, subjecting aryl alkene **50**, bearing a cyclopropyl group at the styrenyl α -position, to the standard reaction conditions with sulfoxonium ylide **2** afforded the radical ring-opened product **51** in 31% yield (see Supporting Information for its SCXRD analysis), while no product retaining the cyclopropyl group (**52**) was observed (Scheme 4c). This outcome suggests that the reaction will proceed via selective addition of a nickel carbene radical to the styrenyl β -position, generating a benzylic radical intermediate that, in this case, undergoes cyclopropyl ring-opening, radical cyclization, and aromatization (highlighted in blue box).

To further elucidate the proposed mechanistic steps, we performed control experiments with deuterated reagents (Scheme 4d). Substitution of the hydride source with deuterated DBpin resulted in selective deuterium incorporation solely at the benzylic position of the corresponding ketone product 3-D1. This observation rules out olefin hydronickelation of NiH prior to carbene activation and, therefore, it suggests that the reaction proceeds via an intramolecular metal hydride transfer process. In contrast, using D_2O as the proton source led to exclusive deuterium incorporation at the α -keto position (3-D2), indicative of product release through protodemetalation of the postulated Ni-alkyl intermediate (Int4 in Scheme 4a) in the final step.

CONCLUSION

In summary, we disclose herein the first example of NiH-catalyzed hydroalkylation of aryl alkenes mediated by a nickel carbene radical, achieving the elusive β -selective hydroalkylation with excellent regioselectivity. The key to this transformation is the preferential engagement of the NiH species with sulfoxonuim ylides to generate a nickel carbene radical, which selectively adds to the styrenyl β -position of the aryl alkene, followed by intramolecular metal hydride transfer and protodemetalation. Using sulfoxonium ylides as bench-stable carbene precursors, the reaction displays broad applicability to diverse alkenes under mild, operationally simple conditions. Combined experimental and computational studies reveal the key mechanistic features of carbene radical intermediacy underlying this transformation.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/jacs.5c14594.

Experimental procedures, characterization data, spectra for all new compounds, crystallographic data, and Cartesian coordinates of all computed structures (PDF)

Accession Codes

Deposition Numbers 2481786–2481788, 2481792, and 2481804 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via the joint Cambridge Crystallographic Data Centre (CCDC) and Fachinformationszentrum Karlsruhe Access Structures service.

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Author Contributions

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Note

The authors declare no competing financial interest.

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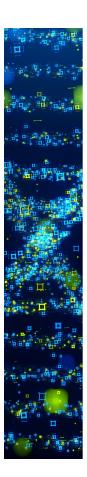
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